

# Pitch-based carbon fibres derived from thermoset fibres oxidized with Cl<sub>2</sub> containing air

T. TOMIOKA\*, Y. ARAI, T. HAMADA†

*Advanced Materials and Technology Research Laboratories, Nippon Steel Corporation, 1618 Ida, Nakahara-ku, Kawasaki, 211 Japan*

Mesophase pitch-based carbon fibres thermoset with Cl<sub>2</sub> containing air were studied for their microstructures and physical properties. Carbon fibres thermoset with Cl<sub>2</sub> containing air and heat-treated at 2000 °C (Cl<sub>2</sub>2000) possessed slightly smaller mean sizes of crystallites  $L_c(002)$ s, lower densities, lower tensile moduli of elasticity, and higher tensile strengths than those thermoset with air. X-ray diffraction measurements revealed a somewhat lower degree of preferred orientation of a carbon fibre which was thermoset with Cl<sub>2</sub> containing air. The Cl<sub>2</sub> thermosetting partly disordered the periodic arrangement of crystallites and reduced the crystallite size  $L_c(002)$  of a carbon fibre heat-treated at a lower temperature. A strong temperature dependence of resistivity was shown for Cl<sub>2</sub>800, suggesting much disorder in its microstructure due to the Cl<sub>2</sub> thermosetting, and Cl<sub>2</sub>1000 and Cl<sub>2</sub>1200, respectively showed specific temperature dependencies of resistivities.

## 1. Introduction

Mesophase pitch-derived precursor fibres are usually thermoset by mild oxidation with air before heat-treatment at higher temperatures [1–3]. The thermoset process converts a thermoplastic pitch precursor fibre infusible and, at the same time, introduces microscopic disorder to the precursor fibre; the degree of preferred orientation of molecules along the fibre axis becomes lower and the molecular stacking height  $L_c(002)$  estimated by X-ray diffraction technique (XRD) gets shorter [4]. The thermosetting with air, however, hardly changes the degree of graphitization of the resultant carbon fibre heat-treated at 2500 °C [5].

Arai *et al.* studied mild oxidation of mesophase pitch-derived pitch precursor fibres with O<sub>2</sub> gas mixed in inert gas, by differential scanning calorimetry (DSC) measurements combined with computer simulation, and proposed a simple model suitable in lower temperature thermosetting such that a pitch precursor fibre possesses a constant number of active sites to react with oxygen, oxygen diffused from the fibre surface reacts with a molecule selectively at the active site, and no further reaction occurs after the all active sites are consumed [6]. They developed the model by assuming two kinds of active sites with different activation energies of oxidation, to give quantitative predictions on the DSC data, the oxygen contents as

functions of the distance from the fibre surface, and the density of fibres thermoset at a variety of conditions of temperature and time [7, 8].

Chlorine gas is an oxidizer and can be used in the thermosetting process, being seeded in oxygen and inert gas mixtures or in air [9]. The radical oxidation by adding Cl<sub>2</sub> gas needs a completely different model for the thermosetting simulation. Preliminary examinations by EPMA (electron probe microanalyser) and DSC measurements combined with computation suggested an almost infinite number of active sites which means a zero-order reaction on the active site number or an oxidation independent of active site number, and very located chlorine atoms in the skin part of the thermoset fibre [9]. The distinctive Cl<sub>2</sub> gas containing thermosetting should also affect the resultant carbon fibres in both their microstructures and physical properties [9].

In this article, density at room temperature ( $d(\text{RT})$ ), tensile modulus of elasticity (TM), tensile strength (TS), interlayer spacing ( $d_{002}$ ) and crystallite size ( $L_c(002)$ ) will characterize the effects of the thermosetting with Cl<sub>2</sub> containing air on pitch-based carbon fibres heat-treated at 2000 °C. Second, detailed XRD measurements will give further information on the microstructures of carbon fibres which have been thermoset with Cl<sub>2</sub> containing air and heat-treated at lower temperatures [10, 11]. Finally, these carbon

\* Present address: Tohoku Kyowa Carbon Corporation, 62-6, Kawauchi, Aza, Nakasoneyama, Osato-cho, Kurokawa-gun, Miyagi, 981-35 Japan

† Present address: Materials Characterization, Technical Development Bureau, Nippon Steel Corporation, 20-1, Shintomi, Futtsu, Chiba, 299-12 Japan

fibres will exhibit different resistivities as functions of temperature from those derived from air-thermoset fibres [12, 13].

## 2. Experimental details

### 2.1. Sample preparation

A mesophase pitch with a softening point of about 304 °C (when the viscosity becomes  $2.0 \times 10^4$  P) and mesophase content of about 94% determined by polarized light microscopy was spun to form pitch precursor fibres. The pitch viscosity during spinning was about  $1.0 \times 10^3$  P and the mean diameter of the spun fibres was about 13  $\mu\text{m}$ .

The pitch precursor fibres were thermoset by mild oxidation in two ways; using dry air or dry air containing 10 mol % of  $\text{Cl}_2$  gas. In thermosetting with air, pitch precursor fibres were heated from 200 °C to 300 °C at a rate of about  $0.5^\circ\text{C min}^{-1}$ , and further treated at 300 °C for 60 min. In the case with 10 mol % of  $\text{Cl}_2$  in air, the fibres were heated at 200 °C for 10 min, temperature was elevated to the final temperature  $T_f$  at a rate of about  $10^\circ\text{C min}^{-1}$ , and the fibres were further treated at  $T_f$  for a treatment period  $\Delta t$ .  $T_f = 260^\circ\text{C}$  and  $300^\circ\text{C}$  and several values of  $\Delta t$  from 0 min to 120 min were tried in the  $\text{Cl}_2$  thermosetting. Thermosetting was carried out with an image furnace; pitch precursor fibres put on a china boat were set in a quartz tube, and heated with infrared light. No fusion among the thermoset fibres was observed after the respective thermosetting.

The thermoset fibres were heat-treated at 2000 °C or at other several temperatures lower than 1400 °C for about 15 min in Ar gas by using a tungsten-heater furnace.

### 2.2. Measurements

The density of a carbon fibre at room temperature ( $d(\text{RT})$ ) was measured by float-sink method; several ethanol-bromoform mixtures with densities differing by  $0.01 \text{ g cm}^{-3}$  were prepared, short-cut fibres were put into test tubes containing these respective liquids at room temperature,  $d(\text{RT})$  of the sample was determined as the mean value of the highest density of the liquid with sinking fibres and the lowest density of the liquid with floating ones. No clear judgement of whether fibres were floating or sinking was regarded to mean that  $d(\text{RT})$  of the fibre tested was equal to that of the liquid.

The tensile modulus of elasticity (TM) and the tensile strength (TS) of single filaments were measured with 25 mm gauge lengths, and about 20 results were averaged for each sample following the standardized method (JIS R 7601-1980) [14]. The filament diameter was measured by using a split image microscope for each filament with an accuracy of about 0.5  $\mu\text{m}$ .

The degree of preferred orientation of crystallites in a fibre was estimated from the half-width at half maximum (HWHM) intensity of the XRD  $\phi$  scan profile, and HWHM was measured as a function of  $2\theta$  from  $18^\circ$  to  $32^\circ$  to investigate the crystallite size ( $L_c(002)$ ) dependence of HWHM in the fibre [10]. Wide angle

XRD profiles ( $\theta$ - $2\theta$  scan profiles) were measured for some fibre rotation angle  $\phi$  to estimate  $L_c(002)$  of the selected crystallites misoriented from the fibre axis by an angle  $\phi$  and to find out whether crystallites in a fibre are periodically arranged or not [10, 11]. Where, the fibre rotation angle,  $\phi$  is the angle between the z-axis and the fibre axis for the experimental set up with the incident X-ray parallel to the x-axis and the X-ray counter in the x-y plane, and  $\phi = 180^\circ$  for the fibre axis parallel to the z-axis and  $\phi = 90^\circ$  for the perpendicular configuration [10]. Resistivity was measured by using the equipment in refs [12] and [13] with single filaments.

In this article, a pitch-based carbon fibre which was thermoset with  $\text{Cl}_2$  containing air and heat-treated at a heat-treatment temperature (HTT) will be noted as  $\text{Cl}_2\text{HTT}$ , and a carbon fibre thermoset by using air and heat-treated at HTT by AirHTT; for example,  $\text{Cl}_21400$ ,  $\text{Cl}_22000$ , or Air2000, where HTT is in °C. The expression of  $\text{Cl}_22000(260, 60)$  means a carbon fibre thermoset using  $\text{Cl}_2$  containing air with  $T_f = 260^\circ\text{C}$  and  $\Delta t = 60$  min and heat-treated at 2000 °C.

## 3. Results and discussion

### 3.1. Density, $d_{002}$ , $L_c(002)$ , and mechanical properties

Fig. 1 shows the density  $d(\text{RT})$  of  $\text{Cl}_22000(300, \Delta t)$  as a function of  $\Delta t$  which is the thermosetting period at the final temperature  $T_f = 300^\circ\text{C}$ . The  $d(\text{RT})$  of Air2000 was about  $2.14 \text{ g cm}^{-3}$ . The  $\text{Cl}_2$  thermosetting effectively decreases the  $d(\text{RT})$  of the carbon fibres heat-treated at 2000 °C more than air thermosetting and a heavier  $\text{Cl}_2$  thermosetting leads to a lower density of the resultant carbon fibre.

No difference in  $d_{002}$  between  $\text{Cl}_2$  thermosetting and one without  $\text{Cl}_2$  gas was detected for the carbon fibres plotted in Fig. 1, the value being about 0.343 nm for both cases.  $L_c(002)$  was slightly depressed by  $\text{Cl}_2$  thermosetting as shown in Fig. 1;  $L_c(002)$  is about 14 nm for Air2000 and it gradually decreases

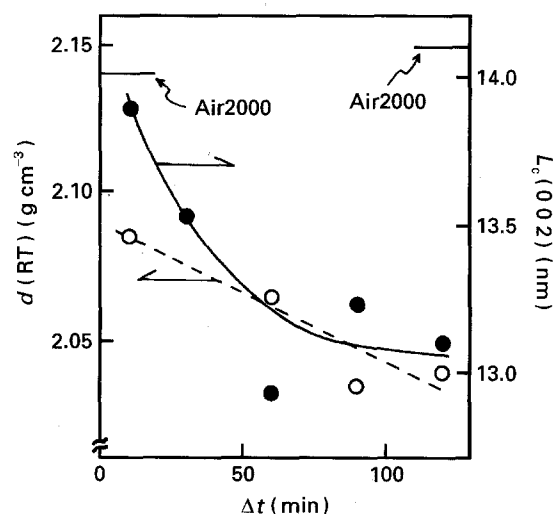


Figure 1 The density at room temperature  $d(\text{RT})$  and the crystallite size  $L_c(002)$  of  $\text{Cl}_22000(300, \Delta t)$  as functions of  $\Delta t$ . The data for Air2000 are also indicated in this figure.

from 14 nm to 13 nm with increase in  $\Delta t$  for  $\text{Cl}_22000(300, \Delta t)$ .

The negligible difference in  $d_{002}$  between  $\text{Cl}_22000(300, \Delta t)$  and Air2000 cannot be evidence of the equal degree of graphitization between them, because  $d_{002}$  exhibits almost equal values for pitch-based carbon fibres heat-treated at 2000 °C in spite of various spinning conditions which result in significant differences in their graphitization [15]. The small difference in  $L_c(002)$  between  $\text{Cl}_22000(300, \Delta t)$  and Air2000 shown in Fig. 1 suggests a weak depression in the degree of graphitization of the resultant carbon fibres heat-treated at 2000 °C [15].

The tensile moduli of elasticity (TM) of  $\text{Cl}_22000(260, \Delta t)$  and  $\text{Cl}_22000(300, \Delta t)$  as functions of  $\Delta t$  are shown in Fig. 2, and the values of Air2000 are also indicated in the figure. The  $\text{Cl}_2$  thermosetting significantly decreases TM for both  $T_f$  values and each  $\Delta t$ , compared to the air thermosetting and a longer  $\text{Cl}_2$  thermosetting results in a lower TM. The increasing TM according to the increasing in  $\Delta t$  for shorter  $\Delta t$  as shown in Fig. 2 may be due to insufficient thermosetting; melting of the core part of a fibre during heat-treatment should misorient the crystallite alignment along the fibre axis.

Fig. 3 shows tensile strength (TS) of the carbon fibres studied in Fig. 2. Fig. 3 indicates that some  $\text{Cl}_2$  thermosetting much improves TS of the carbon fibres for both  $T_f = 260^\circ\text{C}$  and  $300^\circ\text{C}$  and that the condition,  $T_f = 260^\circ\text{C}$ , exhibits a wider range of  $\Delta t$  to improve TS than  $T_f = 300^\circ\text{C}$ . The improved TS and the lowered TM due to the  $\text{Cl}_2$  thermosetting result in a very large strain at tensile break and a toughness improvement of the fibre.

Thus,  $\text{Cl}_2$  thermosetting effectively decreases  $d(\text{RT})$  and TM of pitch-based graphitized fibres, with their degree of graphitization remaining almost constant for a carbon fibre heat-treated at 2000 °C, in contrast

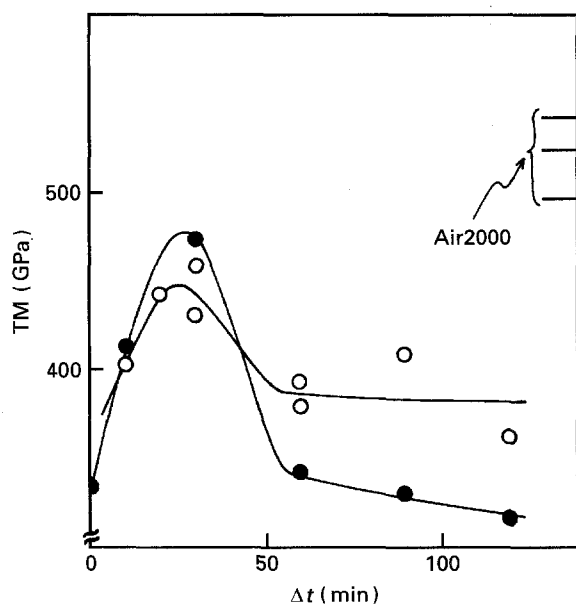


Figure 2 The tensile moduli of elasticity (TM) of  $\text{Cl}_22000(260, \Delta t)$  and  $\text{Cl}_22000(300, \Delta t)$  as functions of  $\Delta t$  which is the period of thermosetting at  $T_f$ . The data for Air2000 are also indicated in this figure.  $\circ$ ,  $T_f = 260^\circ\text{C}$ ;  $\bullet$ ,  $T_f = 300^\circ\text{C}$ .

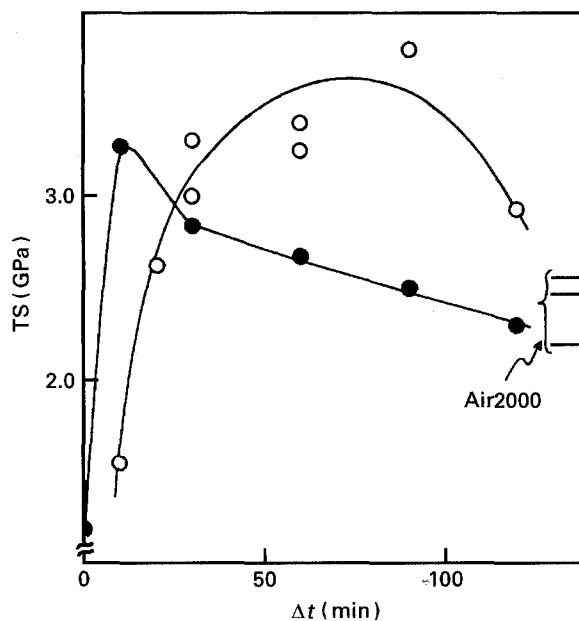


Figure 3 The tensile strengths (TS) of  $\text{Cl}_22000(260, \Delta t)$  and  $\text{Cl}_22000(300, \Delta t)$  as functions of  $\Delta t$ . The data for Air2000 are also indicated in this figure.  $\circ$ ,  $T_f = 260^\circ\text{C}$ ;  $\bullet$ ,  $T_f = 300^\circ\text{C}$ .

to the transverse cross-section controls in spinning previously reported which significantly depress the degree of graphitization of pitch-based graphitized fibres, while keeping the TM constant [5, 16].

A change in TM of a carbon fibre should generally come from an essential structural modification, though only small changes in  $L_c(002)$  of carbon fibres heat-treated at 2000 °C were observed between the two kinds of thermosetting with air and with  $\text{Cl}_2$  containing air. In the next section, degrees of preferred orientation of crystallites along the fibre axis will be investigated for pitch-based carbon fibres thermoset with  $\text{Cl}_2$  containing air and heat-treated at lower temperatures. The preferred orientation of a carbon fibre should directly contribute to its TM [17–19], and the lower heat-treatment temperatures may affect the microstructural differences of carbon fibres between thermosetting with  $\text{Cl}_2$  containing air and with air more than higher temperatures.

### 3.2. Preferred orientation and $\theta$ - $2\theta$ scan profiles

Fig. 4 shows HWHM as functions of  $2\theta$  for  $\text{Cl}_2500(260, 90)$ , and the data of Air500 are also shown in Fig. 4 for comparison. The  $\text{Cl}_2$  thermosetting increases the HWHM of the carbon fibre heat-treated at 500 °C, and the difference in HWHM between  $\text{Cl}_2500(260, 90)$  and Air500 is less dependent on  $2\theta$  from  $18^\circ$  to  $32^\circ$  as shown in Fig. 4. Thus, the treatment reduces the degree of preferred orientation of crystallites in the carbon fibre heat-treated at 500 °C, independent of their size in  $L_c(002)$ , because the data around  $2\theta = 26^\circ$  shown in Fig. 4 mainly come from crystallites with larger  $L_c(002)$  and those far from  $2\theta = 26^\circ$  are primarily from smaller crystallites [10].

Fig. 5 shows HWHMs as functions of  $2\theta$  for  $\text{Cl}_21400(260, 90)$  and Air1400, corresponding to Fig. 4. Heat-treatment at 1400 °C significantly reduces

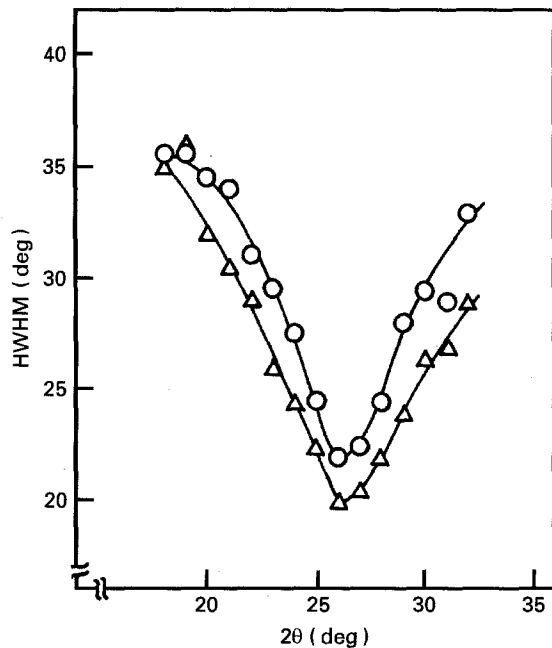


Figure 4 HWHMs as functions of  $2\theta$  for  $\text{Cl}_2 500(260, 90)$  (○) and Air500 (△).

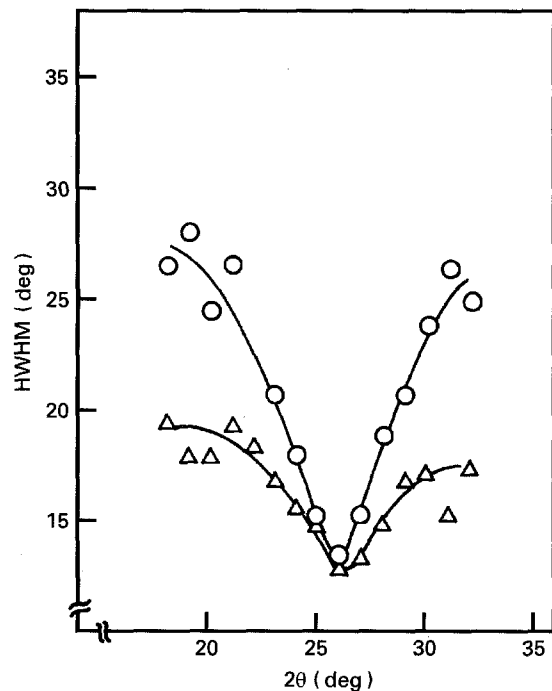


Figure 5 HWHMs as functions of  $2\theta$  for  $\text{Cl}_2 1400(260, 90)$  (○) and Air1400 (△).

the differences in HWHM around  $2\theta = 26^\circ$  between the two kinds of thermosetting observed in the case of  $\text{HTT} = 500^\circ\text{C}$ , but remains or extends the differences for  $2\theta$  far from  $26^\circ$ .

HWHMs at  $2\theta = 26^\circ$  and  $32^\circ$  are plotted as functions of heat-treatment temperature (HTT) in Fig. 6 for both  $\text{Cl}_2$  thermosetting and air thermosetting. HWHM at  $26^\circ$  of  $2\theta$  is mainly due to crystallites with larger  $L_c(002)$  and that at  $32^\circ$  comes from smaller crystallites in  $L_c(002)$  as mentioned above. Fig. 6 clearly shows that the difference in HWHM of larger crystallites between  $\text{Cl}_2$  thermosetting and air thermosetting is vanishing by higher temperature heat-

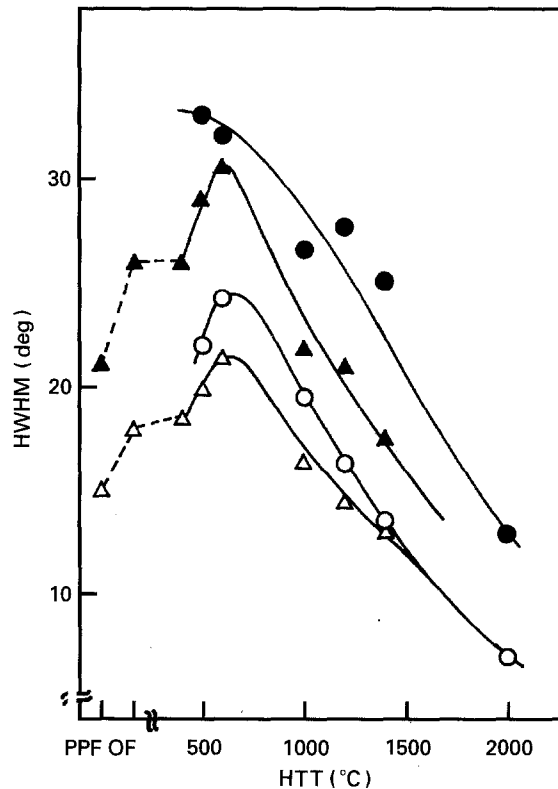


Figure 6 HWHMs at  $2\theta = 26^\circ$  and  $32^\circ$  for two kinds of carbon fibres which were thermoset using  $\text{Cl}_2$  containing air at  $T_f = 260^\circ\text{C}$  and  $\Delta t = 90$  min or with air and heat-treated at various temperatures. HWHMs are plotted as functions of heat-treatment temperature and the values of HWHM of a pitch precursor fibre and a thermoset fibre oxidized with air are also plotted in this figure. ○,  $\text{Cl}_2$  ( $26^\circ$ ); ●,  $\text{Cl}_2$  ( $32^\circ$ ); △, air ( $26^\circ$ ); ▲, air ( $32^\circ$ ).

treatments, but the difference for smaller crystallites remains after heat-treatment at higher temperatures. The low degree of preferred orientation of smaller crystallites may contribute the depressed TM of  $\text{Cl}_2 2000$  which has been mentioned in Fig. 2.

Figs 7 and 8 show  $\theta$ - $2\theta$  scan profiles as functions of  $\phi$  from  $150^\circ$  to  $180^\circ$  for  $\text{Cl}_2 500(260, 90)$  and Air500 respectively. Air500 exhibits wide peaks around  $2\theta = 7^\circ$  corresponding to a periodic arrangement of crystallites [10, 11], but the peaks disappear for  $\text{Cl}_2 500$ .  $\text{Cl}_2 500$  showed unusual  $\phi$  scan profiles for  $2\theta = 4^\circ, 6^\circ$ , and  $8^\circ$  which have a minimum at  $\phi = 180^\circ$ , and the  $\text{Cl}_2$  thermosetting would partly destroy the periodic structure in crystallite arrangement.

Crystallite sizes  $L_c(002)$ s were calculated from the widths of the  $\theta$ - $2\theta$  scan profiles shown in Figs 7 and 8 and are plotted as functions of  $\phi$  for  $\text{Cl}_2 500$  and Air500 in Fig. 9. Fig. 9 shows that a misaligned crystallite possesses a little smaller  $L_c(002)$  than a well-aligned one for both cases [10], and that  $\text{Cl}_2$  thermosetting slightly depressed the size  $L_c(002)$  of a pitch-based carbon fibre heat-treated at  $500^\circ\text{C}$ , compared to the air thermosetting.

### 3.3. Electrical resistivity

Electrical resistivity  $\rho$  of  $\text{Cl}_2 800(260, 90)$  is shown in Fig. 10. The figure shows that the resistivity monotonously decreases with increasing temperature from 3 K to 300 K; the resistivity at 4 K,  $\rho(4)$  is about

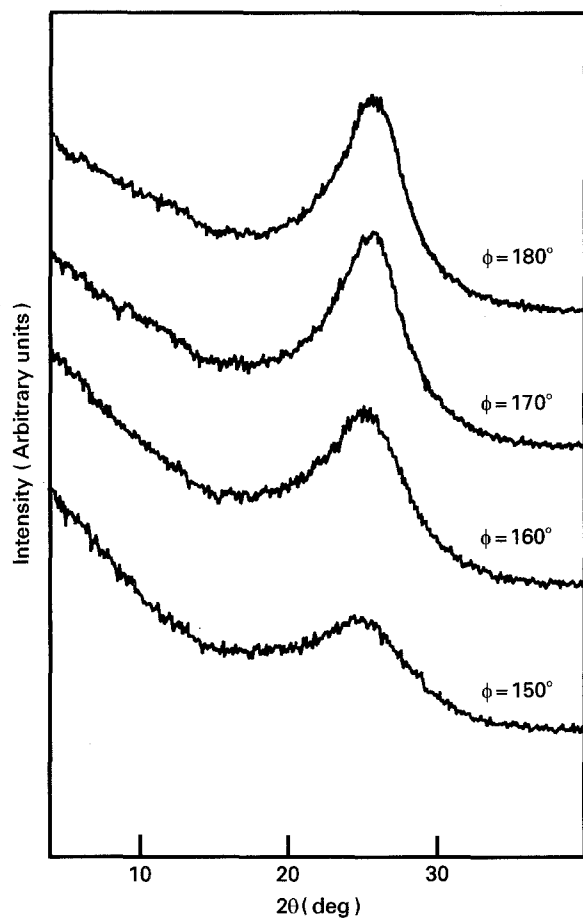


Figure 7 The  $\theta$ - $2\theta$  scan profiles of Cl<sub>2</sub>500(260, 90) for  $\phi = 180^\circ$ ,  $170^\circ$ ,  $160^\circ$ , and  $150^\circ$ .

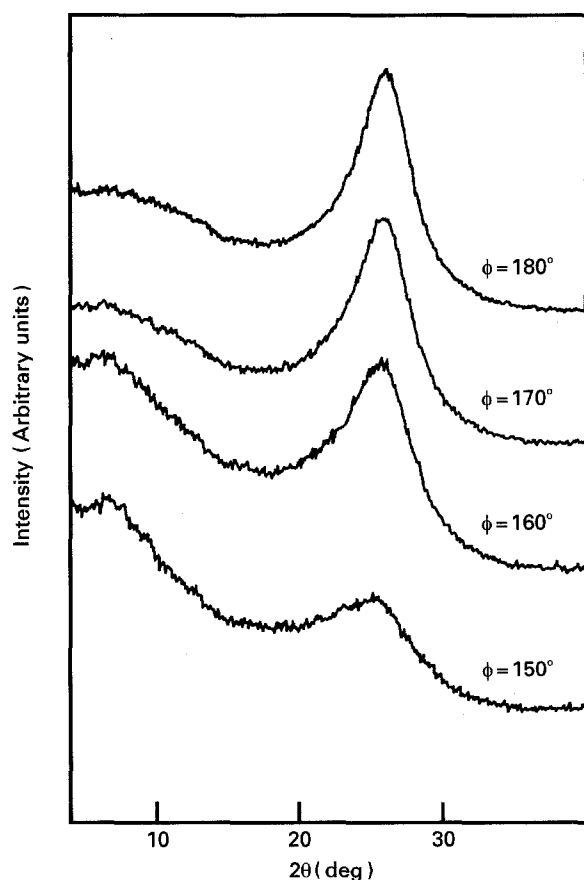


Figure 8 The  $\theta$ - $2\theta$  scan profiles of Air500 for  $\phi = 180^\circ$ ,  $170^\circ$ ,  $160^\circ$ , and  $150^\circ$ .

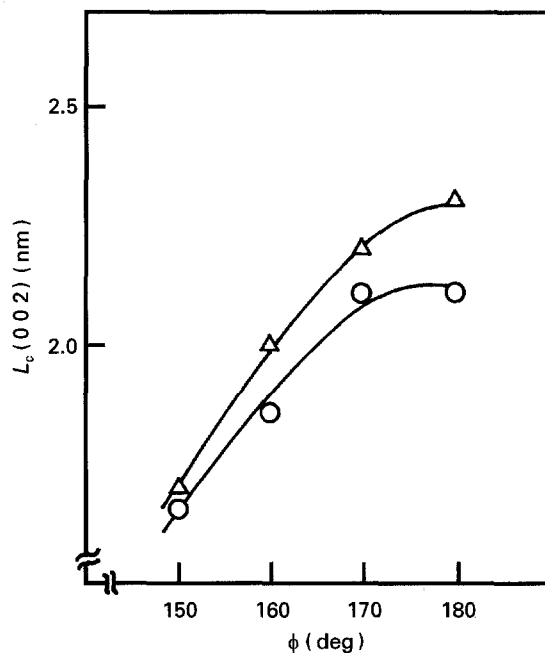


Figure 9 The crystallite size,  $L_c(002)$ , of Cl<sub>2</sub>500(260, 90) (○) and Air500 (△) as functions of  $\phi$ .

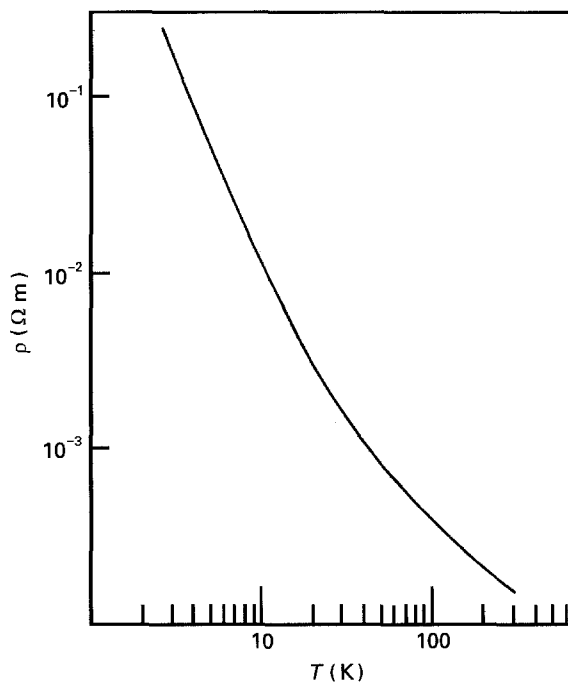


Figure 10 The temperature dependence of the electrical resistivity,  $\rho$ , of Cl<sub>2</sub>800(260, 90).

$5.3 \times 10^2$  times larger than that at 300 K,  $\rho(300)$ . On the other hand, Air800 exhibited a monotonous and much weaker temperature dependence of  $\rho$ , the ratio  $\rho(4)/\rho(300)$  was about 1.9. The strong temperature dependence shown in Fig. 10 of Cl<sub>2</sub>800 corresponds to a non-metallic electrical conductivity and suggests that Cl<sub>2</sub>800 is a more disordered system than Air800 [20–24].

The ratio  $\rho(4)/\rho(300)$  of a pitch-based carbon fibre heat-treated at lower temperatures is less affected by a transverse cross-section control in spinning [13]: Air800 with a diameter of about 10  $\mu m$  and one heat-treated at 800 °C with a 25  $\mu m$  diameter

which was noted as #A800 and investigated in ref. [13] showed almost the same value of  $\rho(4)/\rho(300)$ , and Air1000 and Air1200 respectively showed almost identical temperature dependencies of  $\rho$  with corresponding #A1000 and #A1200 having about 25  $\mu\text{m}$  of diameters studied in refs [12] and [13], where #A800, #A1000 and #A1200 were thermoset by using air at the same conditions with Air800. Thus, the  $\text{Cl}_2$  thermosetting would change the microstructures of a carbon fibre in a shorter scale than the cross-section control in spinning related to a scale of several tens of nanometres [5, 25].

$\text{Cl}_2$ 800 showed a stronger temperature dependence of  $\rho$  than Air700 with a rate  $\rho(4)/\rho(300)$  of about  $6.9 \times 10^1$  and a weaker one than a carbon fibre (#A600) thermoset by using air at the same conditions with Air800 and heat-treated at 600 °C with a diameter of about 25  $\mu\text{m}$  which was studied in [13] as #A800, #A1000 and #A1200 mentioned above. The ratio  $\rho(4)/\rho(300)$  suggests that  $\text{Cl}_2$ 800 should correspond to an air-thermoset carbon fibre heat-treated at a temperature between 600 °C and 700 °C.

The resistivities of  $\text{Cl}_2$ 1000 and  $\text{Cl}_2$ 1200 are shown in Fig. 11 and the detailed temperature dependence of  $\rho$  at lower temperatures is exhibited in Fig. 12 for  $\text{Cl}_2$ 1200. The  $\rho$  of  $\text{Cl}_2$ 1000 monotonously decreases

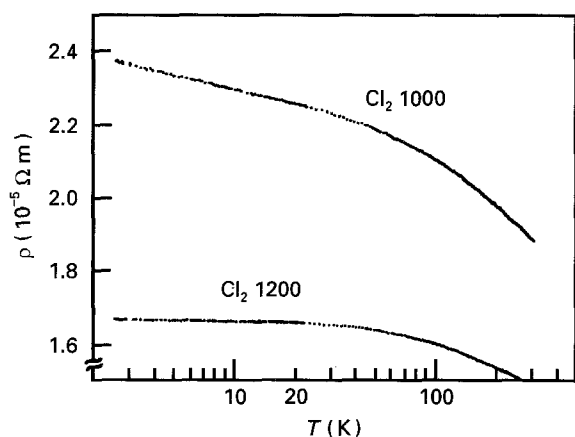


Figure 11 The temperature dependence of the electrical resistivities,  $\rho$ , of  $\text{Cl}_2$ 1000(260, 90) and  $\text{Cl}_2$ 1200(260, 90).

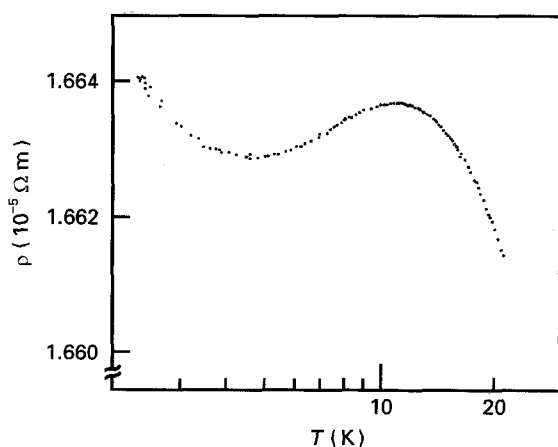


Figure 12 The electrical resistivity,  $\rho$ , of  $\text{Cl}_2$ 1200(260, 90) as a function of temperature from about 3 K to about 30 K.

with increasing temperature, although Air1000 and #A1000 had a maximum around  $T = 20$  K, and the temperature dependence of  $\text{Cl}_2$ 1000 was significantly stronger than those of Air1000 or #A1000 [12, 13]. The stronger temperature dependence of  $\text{Cl}_2$ 1000 may inherit the resistivity of  $\text{Cl}_2$ 800 which was more disordered and non-metallic.

$\text{Cl}_2$ 1200 shows a very weak temperature dependence of  $\rho$ , especially in the lower temperature region and possesses a maximum at about  $T = 12$  K and minimum at about  $T = 5$  K as shown in Figs 11 and 12. Air1200 and #A1200 exhibited a maximum at about  $T = 20$  K and no minimum for  $T$  higher than 2 K, similar to Air1000 or #A1000 [12, 13].

Koike and Fukase studied resistivities of carbon fibres derived from pitch and heat-treated at 1200 °C [26]. They showed that the temperature dependencies of the carbon fibres at temperatures below 10 K under a magnetic field or without magnetic field can be quantitatively explained by electron-electron interaction [27, 28], and the Kondo effect in a disordered three-dimensional system [29, 30]. A strong magnetic field applied to the carbon fibre depressed the Kondo effect and shifts a minimum position in  $\rho$  as the function of temperature to higher temperatures [26]; minimum at about  $T = 1.5$  K without magnetic field and minimum at about  $T = 5$  K under a magnetic field of 56 kOe. Thus, the minimum of  $\rho$  at about  $T = 5$  K of  $\text{Cl}_2$ 1200 might be due to the small Kondo effect. The increase in  $\rho$  with increasing temperature from about 5 K to about 10 K would be due to electron-phonon scattering [26], and a small electron-phonon scattering may be another explanation of the minimum at the higher temperature and the maximum at the lower temperature of  $\text{Cl}_2$ 1200.

#### 4. Conclusions

Thermosetting using  $\text{Cl}_2$  containing air caused a lower density, a slightly depressed graphitization, a significantly lower tensile modulus of elasticity, and a much improved tensile strength of resultant carbon fibres heat-treated at 2000 °C. The lower tensile modulus of elasticity with an almost constant degree of graphitization by the  $\text{Cl}_2$  thermosetting makes contrast with a constant tensile modulus and significantly lowered graphitization induced by transverse cross-section control in spinning.

The  $\text{Cl}_2$  thermosetting misaligned crystallite orientation along the carbon fibre axis for a carbon fibre heat-treated at 500 °C, independent on the crystallite size  $L_c(002)$ . The misoriented crystallites with smaller sizes were less easily well-aligned than larger ones by further heat-treatment at temperatures higher than 1400 °C. The small crystallites which remained less oriented would partly contribute to the lower tensile modulus of elasticity of a carbon fibre thermoset with  $\text{Cl}_2$  containing air and heat-treated at higher temperatures.

The  $\text{Cl}_2$  thermosetting partially destroyed the periodic arrangement of crystallites and reduced the crystallite size  $L_c(002)$  of a carbon fibre heat-treated at 500 °C.

The Cl<sub>2</sub> thermosetting introduced microstructural disorders to a carbon fibre heat-treated at 800 °C, and the resistivity,  $\rho$ , showed a strong temperature dependence. A carbon fibre thermoset with Cl<sub>2</sub> containing air and heat-treated at 1000 °C had a monotonously decreasing  $\rho$  with increasing temperature, and one Cl<sub>2</sub> thermoset and heat-treated at 1200 °C exhibited  $\rho$  as a function of temperature with a maximum at  $T = 12$  K and a minimum at  $T = 5$  K, while both carbon fibres thermoset with air and heat-treated at 1000 °C and 1200 °C showed a maximum at  $T = 20$  K and no minimum in the temperature range higher than 2 K.

## References

1. D. D. EDIE, N. K. FOX, B. C. BARNETT and C. C. FAIN, *Carbon* **24**(4) (1986) 477.
2. A. A. BRIGHT and L. S. SINGER, *Carbon* **17** (1979) 59.
3. S. OTANI, *Mol. Cryst. Liq. Cryst.* **63** (1981) 249.
4. D. S. KURTZ, MSc thesis, Rensselaer Polytechnic Institute, Troy, New York, USA (1983).
5. T. HAMADA, T. NISHIDA, M. FURUYAMA and T. TOMIOKA, *Carbon* **26**(6) (1988) 837.
6. Y. ARAI, T. IWASHITA and T. TOMIOKA, Preprint for the twenty-second autumn conference of Chemical Engineering Society of Japan, Tokyo (1989) p. 303 (in Japanese).
7. Y. ARAI, T. IWASHITA and T. TOMIOKA, Preprint for the sixteenth annual meeting of Carbon Society of Japan, Tokyo (1989) p. 174 (in Japanese).
8. Y. ARAI, T. IWASHITA and T. TOMIOKA, Preprint for the International Symposium on Carbon, Tsukuba, Japan (1990) p. 782.
9. Y. ARAI, T. TOMIOKA, Preprint for the eighteenth annual meeting of Carbon Society of Japan, Saitama (1991) p. 76 (in Japanese).
10. T. HAMADA, M. FURUYAMA, T. TOMIOKA and M. ENDO, *J. Mater. Res.* **7**(9) (1992) 2612.
11. T. HAMADA and T. TOMIOKA, *Carbon* **31**(1) (1993) 235.
12. T. HAMADA, K. KAWASAKI, M. FURUYAMA and T. TOMIOKA, *J. Phys. Soc. Japan* **59**(7) (1990) 2468.
13. T. HAMADA, PhD thesis, Tokyo University, Tokyo (1990) (in Japanese).
14. JIS (Japan Industrial Standard) R 7601-1980; Testing Methods for Carbon Fibers (1980) (in Japanese).
15. T. HAMADA, M. FURUYAMA, Y. SAJIKI, T. TOMIOKA and M. ENDO, *J. Mater. Res.* **5**(3) (1990) 570.
16. T. HAMADA, T. NISHIDA, Y. SAJIKI, M. MATSUMOTO and M. ENDO, *J. Mater. Res.* **2**(6) (1987) 850.
17. R. J. PRICE, *Philos. Mag.* **12** (1965) 564.
18. P. R. GOGGIN and W. N. REYNOLDS, *Philos. Mag.* **16** (1967) 317.
19. M. S. DRESSELHAUS, G. D. DRESSELHAUS, K. SUGIHARA, I. L. SPAIN and H. A. GOLDBERG, "Graphite Fibers and Filaments", edited by M. Gardona (Springer Series in Materials Science 5, 1988) p. 92 and p. 132.
20. S. MROZOWSKI, *Carbon* **9** (1971) 97.
21. H. B. BROWN, Y. TOMKIEWICZ, A. A. AVIRAM, A. BROERS and B. SUNNERS, *Solid State Comm.* **35** (1980) 135.
22. W. BUCKER, *J. Non-Cryst. Solids* **12** (1973) 115.
23. I. L. SPAIN, K. J. VOLIN, H. A. GOLDBERG and I. KALNIN, *J. Phys. Chem. Solids* **44** (1983) 839.
24. B. DEROIDE, J. V. ZANCHETTA and A. DIBY, *Carbon* **29**(1) (1991) 3.
25. T. HAMADA, Y. SAJIKI, M. FURUYAMA, T. TOMIOKA and M. ENDO, *J. Mater. Res.* **4**(4) (1989) 1027.
26. Y. KOIKE and T. FUKASE, *Solid State Comm.* **62**(7) (1987) 499.
27. Y. ISAWA, K. HOSHINO and H. FUKUYAMA, *J. Phys. Soc. Japan* **51** (1982) 3262.
28. Y. ISAWA and H. FUKUYAMA, *J. Phys. Soc. Japan* **53** (1984) 1415.
29. F. J. OHKAWA and H. FUKUYAMA, *J. Phys. Soc. Japan* **53** (1984) 2640.
30. F. J. OHKAWA, *Prog. Theor. Phys. Suppl.* **84** (1985) 166.

Received 2 July 1993  
and accepted 8 September 1994